

Figure 1a: C. Adachi, et. al.

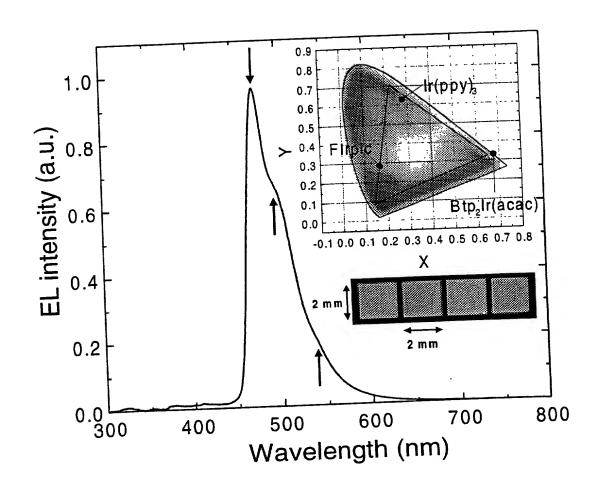


Figure 1b: C. Adachi, et. al.

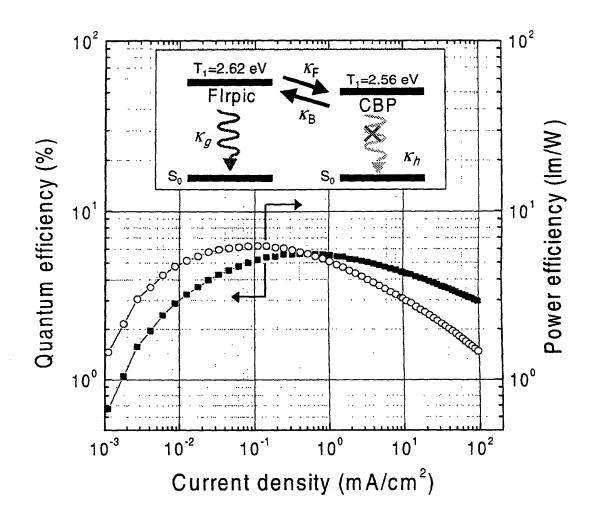


Figure 2: C. Adachi, et.al.

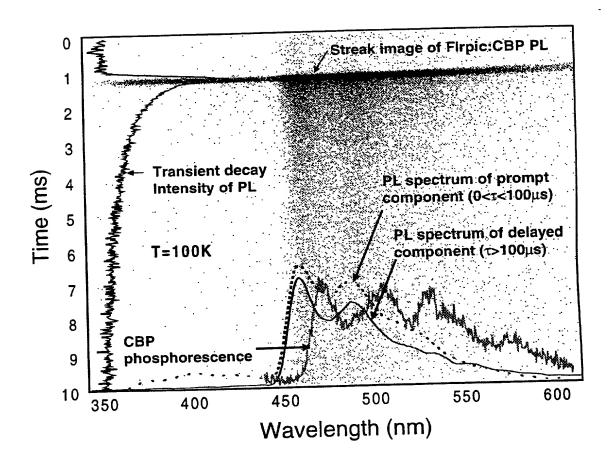


Figure 3: C. Adachi, et. al.

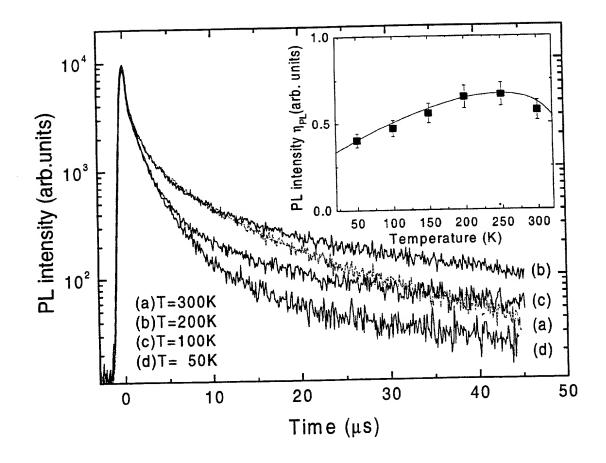


Figure 4: C. Adachi, et.al.

Generic Mono-Anionic, Bidentate, Carbon-Coordination Ligands-I

X = S, O, NR; and R_1 , R_2 , R_3 , R_4 and R_5 are, independently, hydrogen, halogen, alkyl, aryl or arylene; and R'_1 and R'_2 may, in combination, be aryl.

Generic Mono-Anionic, Bidentate, Carbon-Coordination Ligands-II

X = S, O, NR; and R_1 , R_2 , R_3 , R_4 and R_5 are, independently, hydrogen, halogen, alkyl, aryl or arylene; and R'_1 and R'_2 may, in combination, be aryl.

Generic Mono-Anionic, Bidentate, Carbon-Coordination Ligands-III

$$R_1$$
 S
 N
 R_2

$$R_3$$
 R_2

X = S, O, NR; and R_1 , R_2 , R_3 , R_4 and R_5 are, independently, hydrogen, halogen, alkyl, aryl or arylene.

Figure 5d

Specific Mono-Anionic, Bidentate, Carbon-Coordination Ligands-I

Figure Eq

Generic Non-Mono-Anionic, Bidentate, Carbon-Coordination Ligands-I

R, R₁, R₂, R₃, R₄, R₅, and R₆ are, independently, hydrogen, halogen, alkyl or aryl.

Generic Non-Mono-Anionic, Bidentate, Carbon-Coordination Ligands-II

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{4}$$

$$R_{2}$$

$$R_{4}$$

$$R_{2}$$

$$R_{3}$$

$$R_{4}$$

$$R_{2}$$

$$R_{4}$$

$$R_{2}$$

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{4}$$

$$R_{4}$$

$$R_{4}$$

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$$R_{3}$$

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$$R_{2}$$

$$R_{3}$$

$$R_{4}$$

$$R_{4}$$

$$R_{2}$$

$$R_{3}$$

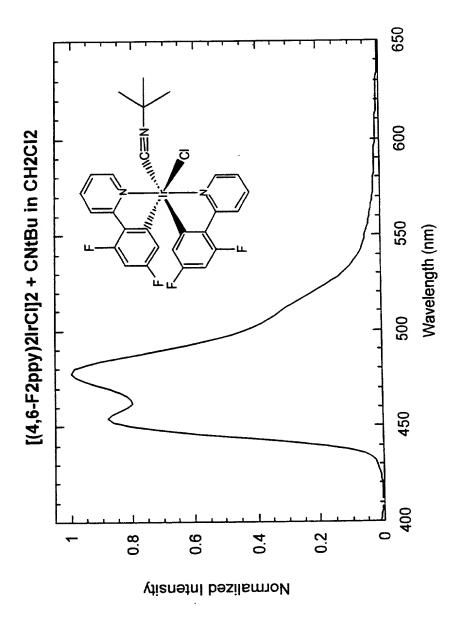
$$R_{4}$$

$$R_{4}$$

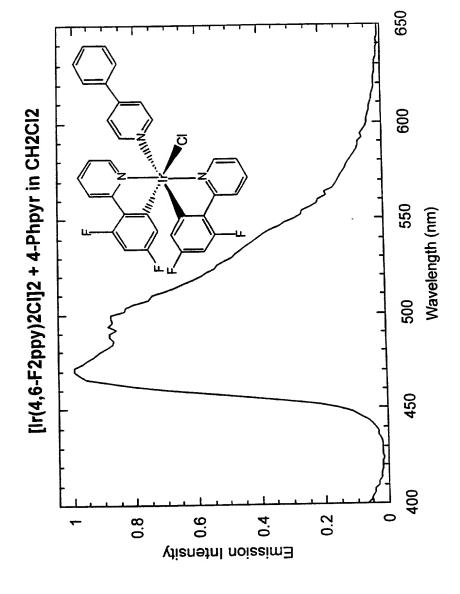
R, R₁, R₂, R₃, R₄, R₅, R₆, R₇ and R₈ are, independently, hydrogen, halogen, alkyl or aryl.

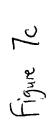
Figure 6c

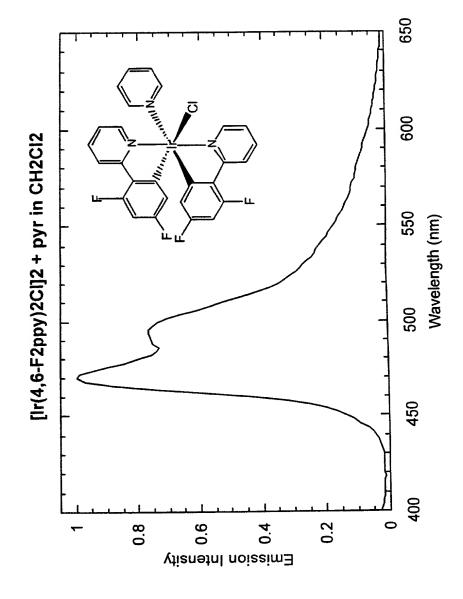
Specific Non-Mono-Anionic, Bidentate, Carbon-Coordination Ligands

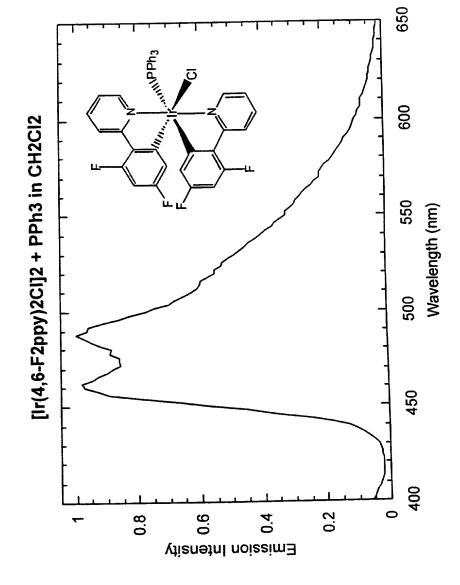


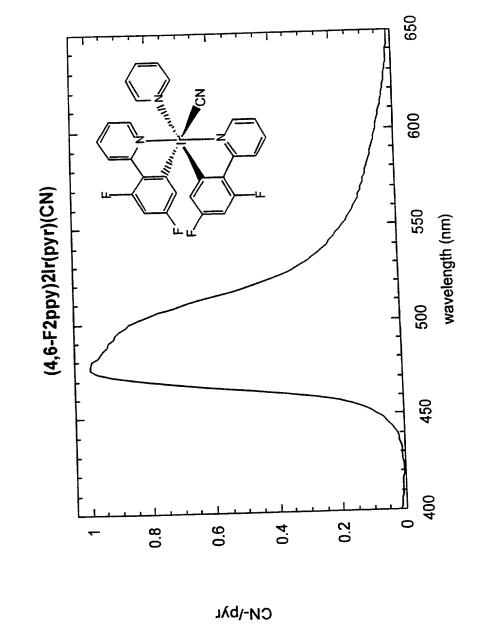


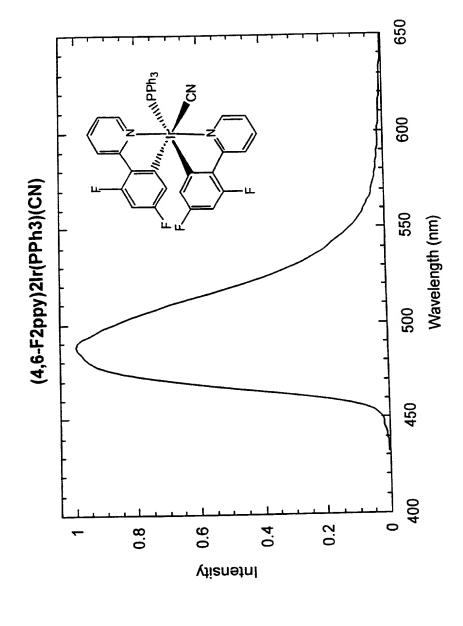


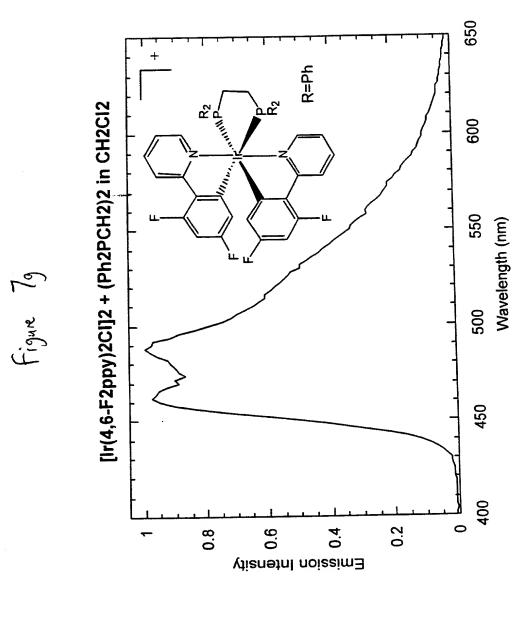


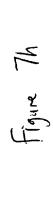


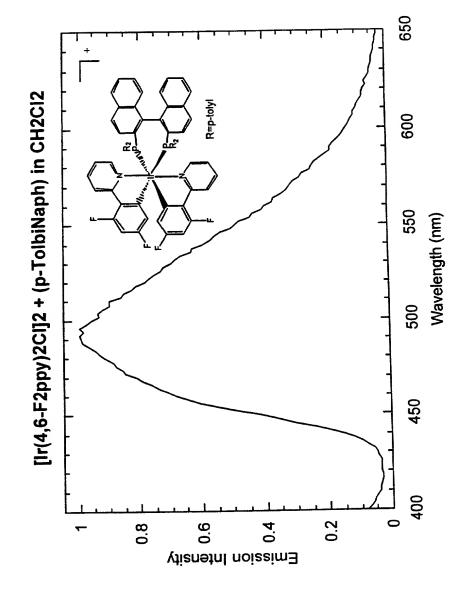


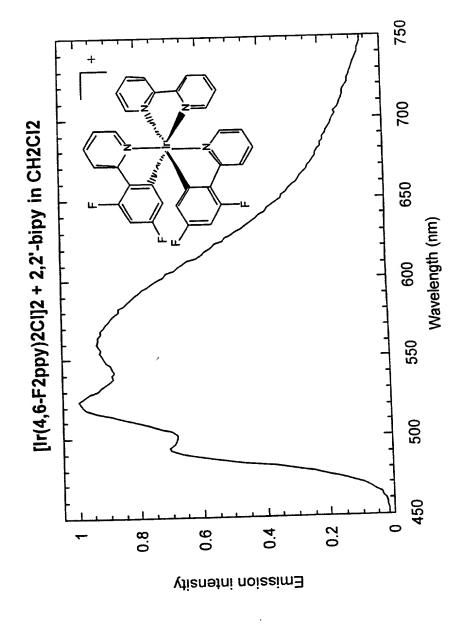


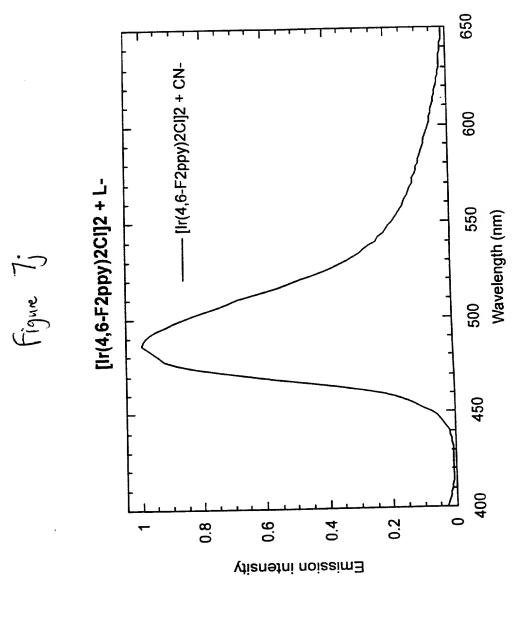


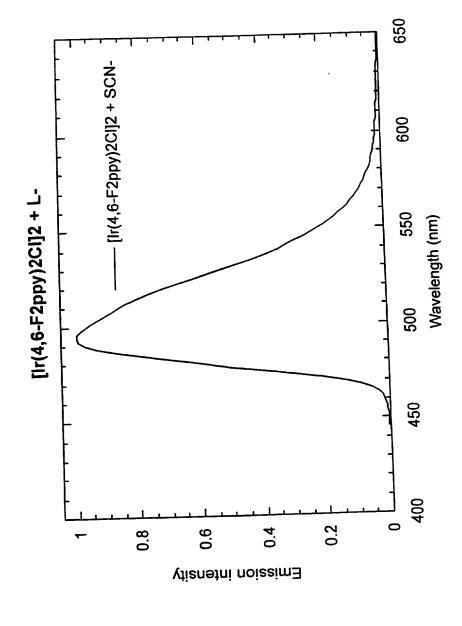


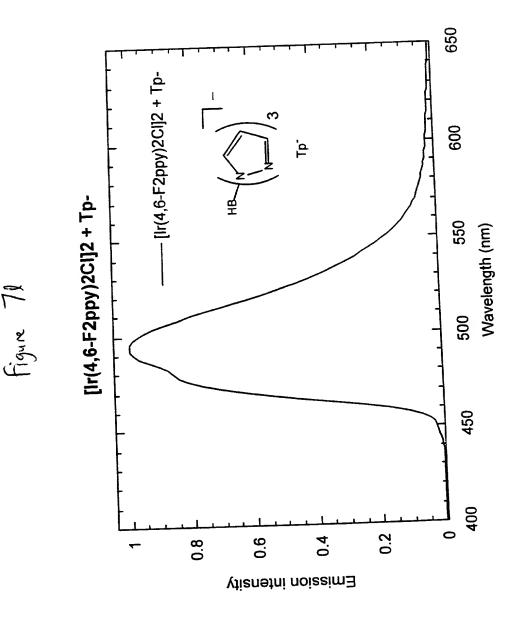


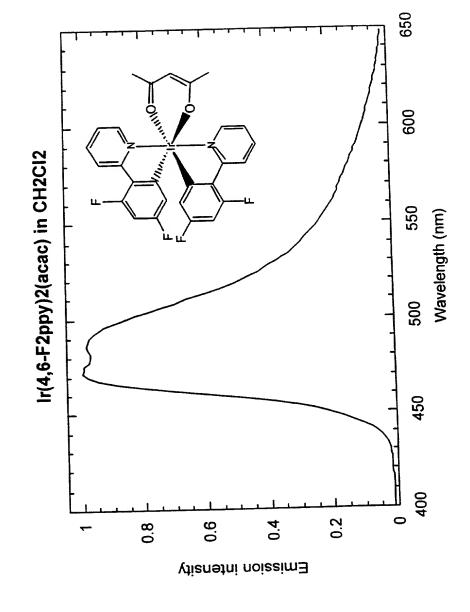


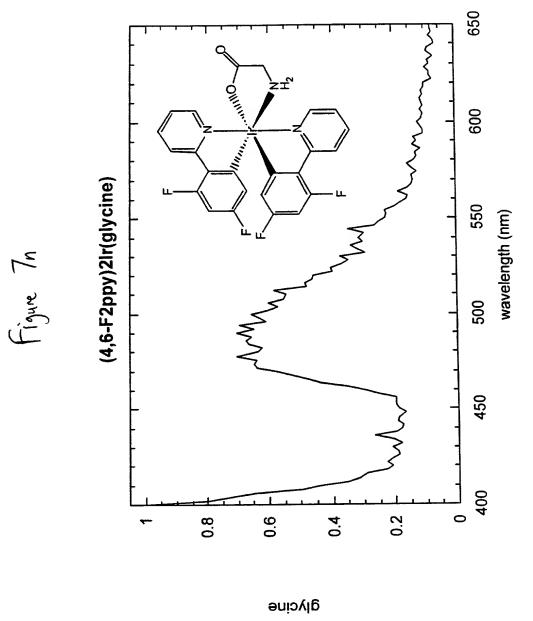


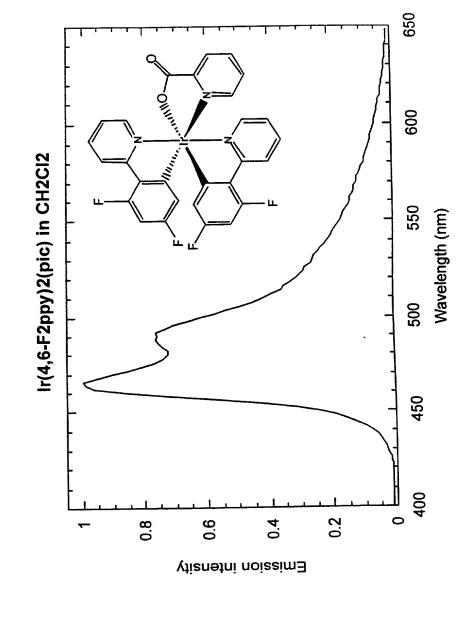




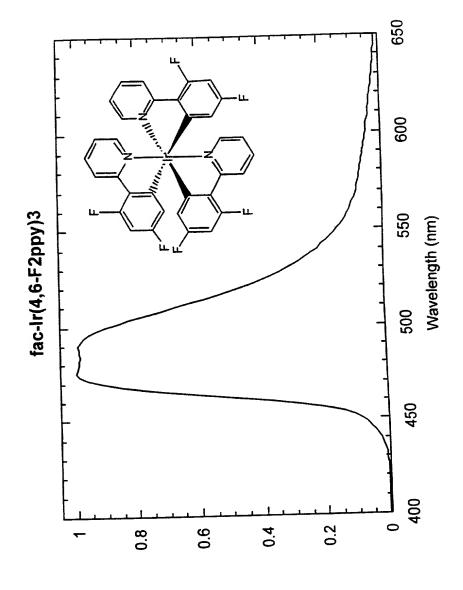




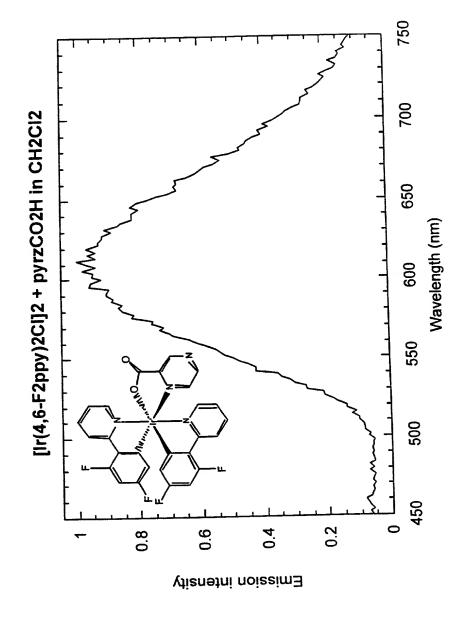








F2irppy em



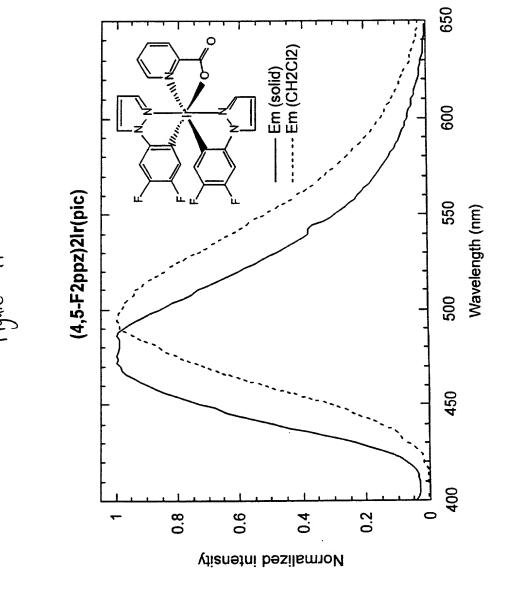


Figure 8a

$$\mathbb{R}_{\mathbb{R}_{2}}^{\mathbb{R}}$$

$$R_{2}P$$
 PR_{2}

X=CH, N E=O,S,Se,Te

Figures 9@> - 9(9)

Figure 10:This Emission spectrum shows the spectra of both $Pt(ppy)_2$ and $Pt(ppy)_2Br_2$. The former gives green emission, partly form MLCT transitions, and the latter gives blue emission, predominantly from a triplet π - π * transition. The structure observed for the $Pt(ppy)_2Br_2$ spectrum is consistent with ligand centered emission. The luminescent lifetimes for the two complexes are 4 and 150 µsec.

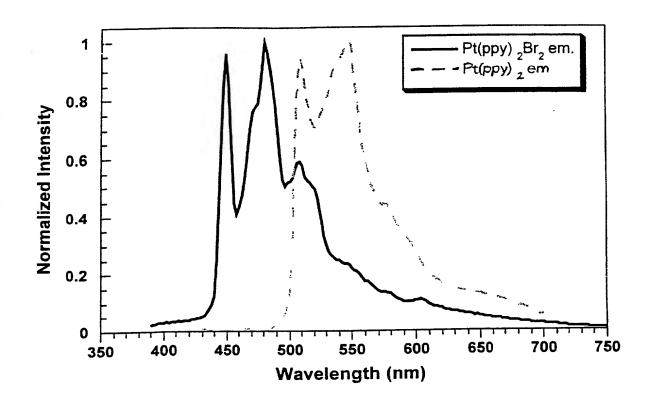


Figure 10

Figure II: This plot shows the emission spectra of (ppy)AuCl₂ and (ppy)A = 2.2'-biphenylene). Both emit from ligand triplet π - π * transitions.

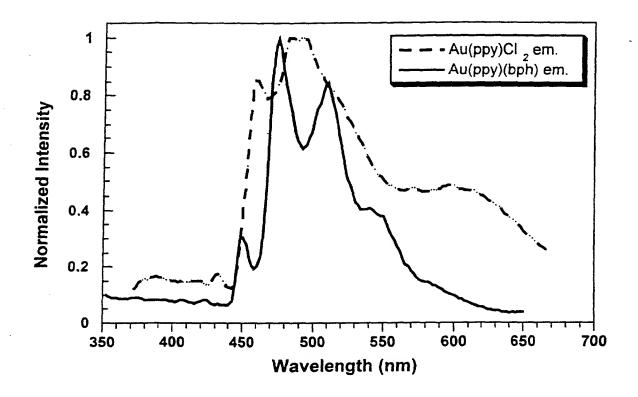


Figure 11

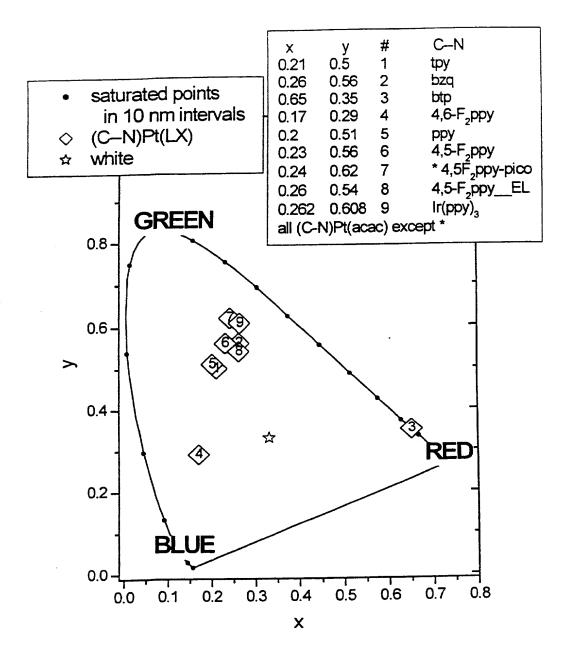


Figure 12

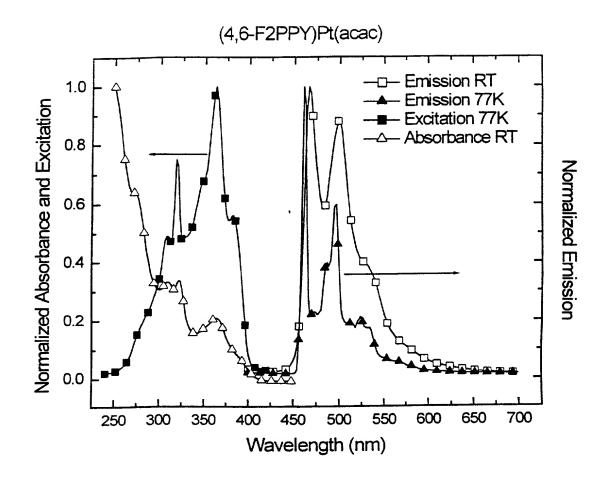


Figure 13

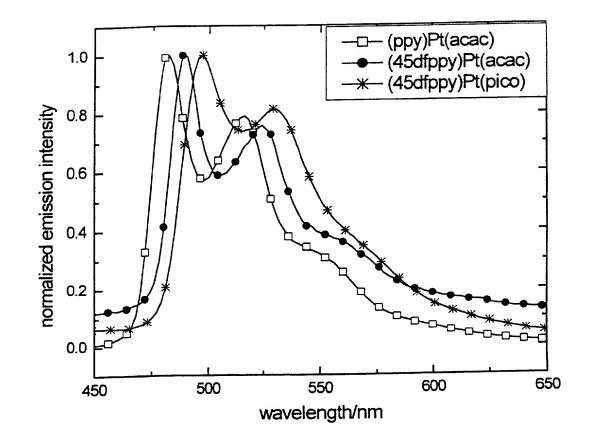


Figure 14

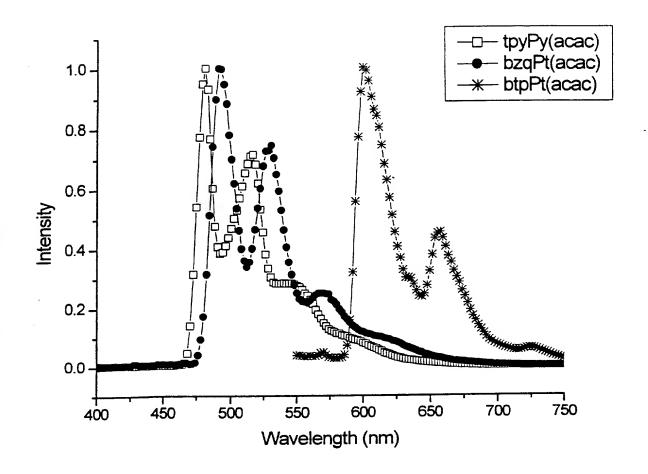


Figure 15

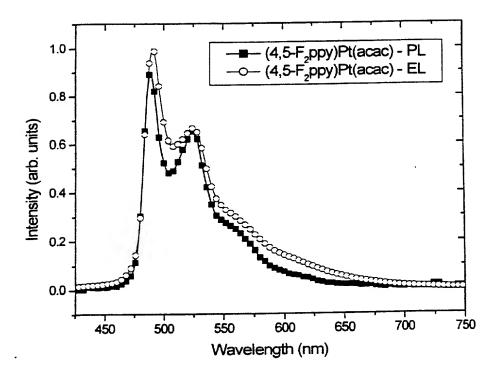


Figure 16

FIG. 17

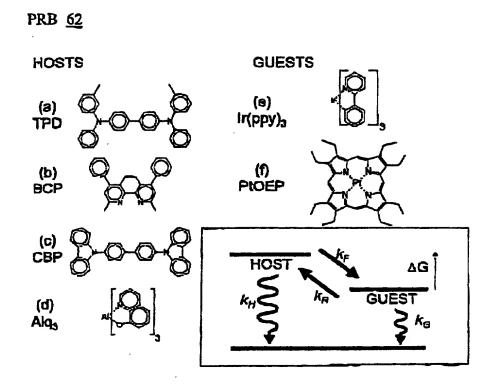


FIG. 18

PRB <u>62</u>

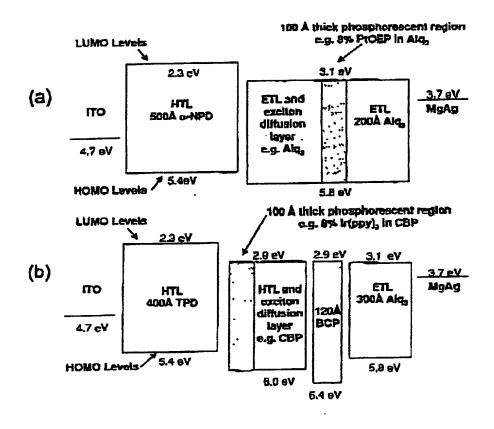


FIG. 19

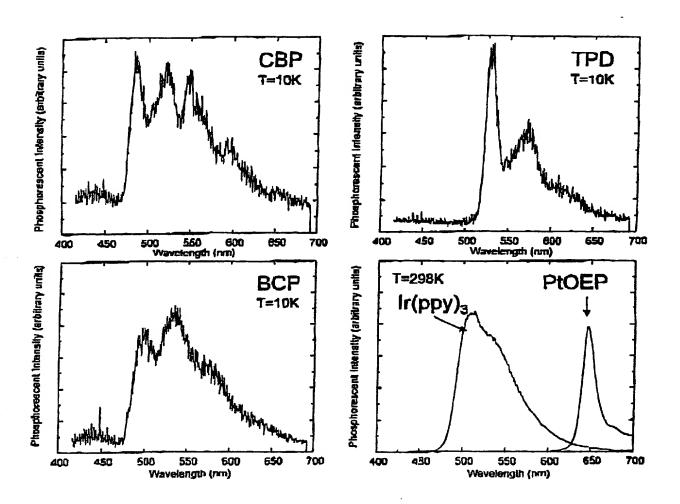


FIG. 20

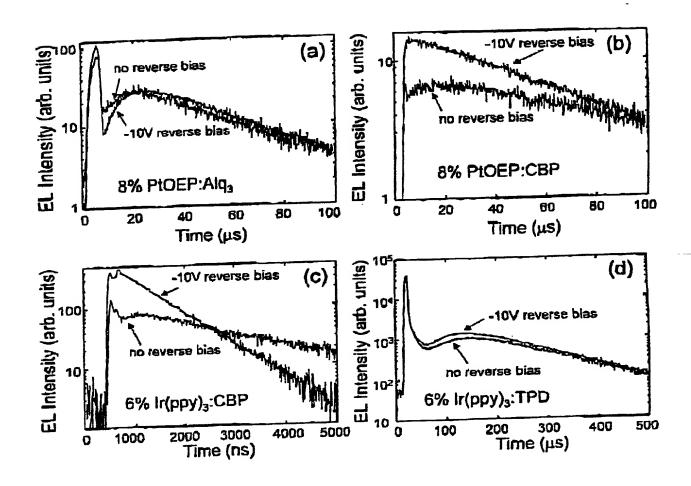


FIG. 21

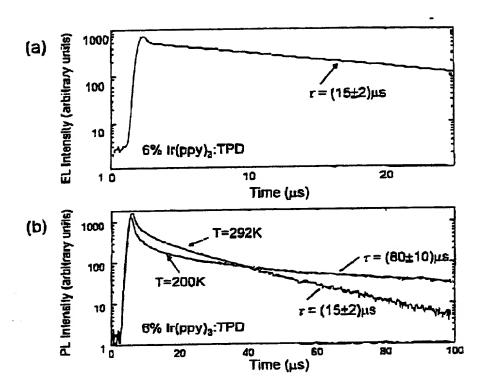


FIG. 22

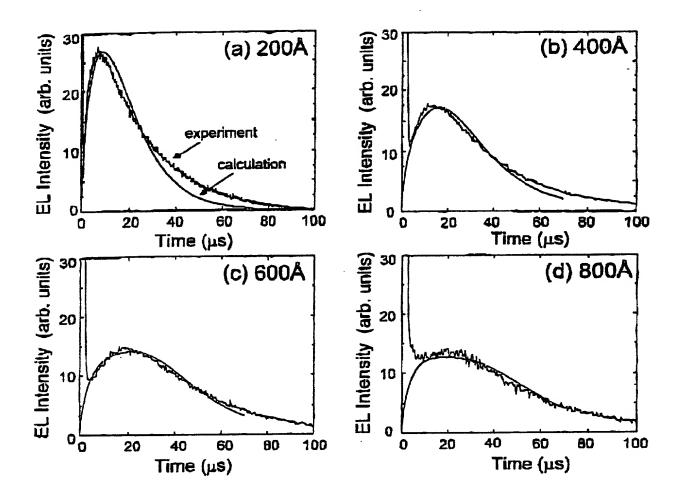


Fig. 23

